

200 L  $F^{3^+} = 0.001$ L<sup>1</sup>. 2 10<sup>5</sup>  $L^1$ 0.001 L¹ 10 L 500 L 10 L 1 L<sup>1</sup> 500 L L<sup>1</sup> F <sup>3+</sup> 50 L 0.001 20, 30, 40 4, 6, 8 10 10 <sup>5</sup> L¹ , (NB: 500 L A ( : ).

500 L .) Preparation of 1 mol L<sup>-1</sup> ammonium thiocyanate

## Preparation of 1 mol L<sup>-1</sup> ammonium thiocyanate solution

38 500 L

## Preparation of 0.15 mol L<sup>-1</sup> potassium permanganate solution

. 2.4 100 L

## Preparation of iron tablet for analysis

1. 100 L 20 L 1 L<sup>1</sup> (see safety notes). A '

. (NB: ) 2. , 0.15 L<sup>1</sup> , .1 , F<sup>2+</sup> F<sup>2+</sup> F<sup>2+</sup> F<sup>3+</sup> .F

2 L 0.15 L<sup>1</sup> . 250 L

## Preparation of food sample for analysis

3.

1. A 2 5 ( ) 2. H ( F 1) ( ) 2), F ( . NB: / . A

. 3.

- . A
- ( F 4). A
- . , 10 L
- 3. 10 L 1 L<sup>1</sup> , 2
- .
- 4. M . A
- . 5. A 15
- , 490
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